

Time: 3Hours

Marks: 100

- N.B.: (1) All questions are compulsory.
 (2) Figures to the right indicate full marks.
 (3) Use of log table/ non-programmable calculator is allowed.

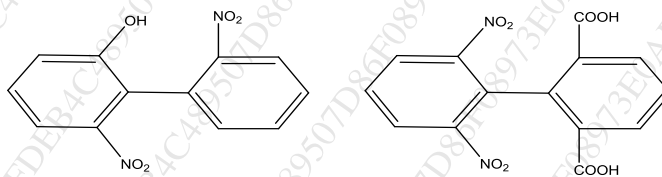
Q.1 Attempt ANY FOUR of the following:

- A) Write short notes on the following: 3
 i) Chelotropic reactions 3
 ii) Electrophilicity 2
- B) Complete the following reaction and name the product. Explain the mechanism of the reaction. 5

$$\text{CH}_3\text{COOH} + \text{C}_2\text{H}_5\text{OH} \xrightleftharpoons{\text{dry HCl gas}} ?$$
- C) Explain the following terms: 5
 i) ligand ii) $\text{B}_{\text{Ac}2}$ iii) pericyclic reaction iv) nucleophile
 v) synartetic acceleration
- D) Using suitable examples explain sigmatropic and group transfer reactions. 5
- E) Distinguish between the following: 3
 i) Fluorescence and Phosphorescence 3
 ii) Singlet and Triplet state 2
- F) What is photosensitisation? Explain the photochemical reduction of benzophenone. 5

Q.2 Attempt ANY FOUR of the following:

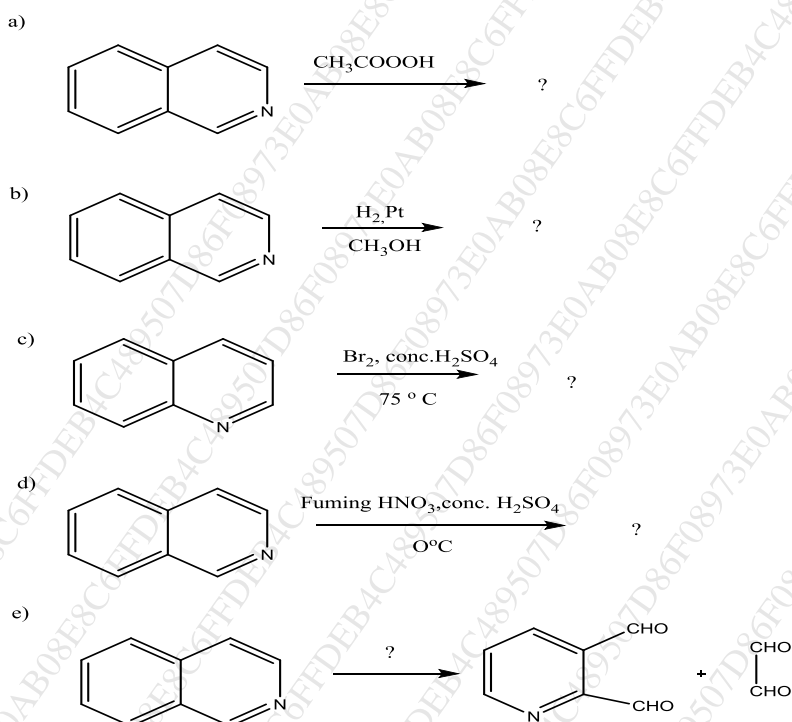
- A) a) State whether following compounds are optically active or optically inactive. Justify your answer. 3



- i) 2
- ii) 2
- b) Define centre of symmetry with an example.
- B) Write a note on stereochemistry of allenes. 5
- C) a) Give synthesis of indole-3-acetic acid. 3
 b) Give preparation of pyridine-N-oxide from pyridine. 2

D) Complete the following reactions.

5



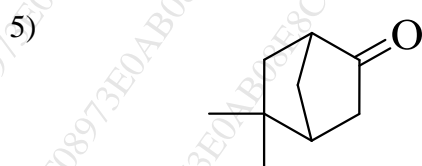
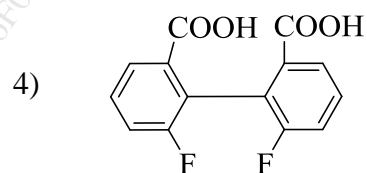
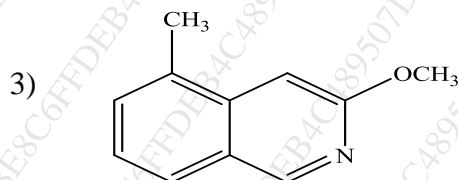
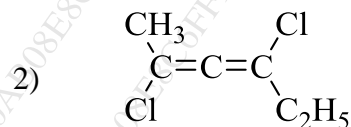
E) What are Agrochemicals? Give advantages of using agrochemicals. Write the synthesis of Endosulfan.

F) Give the Bischler-Napieralski synthesis for the preparation of Isoquinoline. Write the reaction of isoquinoline with alkaline KMnO_4 .

Q.3. Attempt ANY FOUR of the following:

A) Give the IUPAC names of the following:

5



B) Write the structural formula for each of the following compounds: 5

1. 1-chloro-6-methoxyspiro [3.4] octane
2. Bicyclo [3.3.0] octan-3-carboxylic acid
3. 2,2'-difluoro-6,6'-dinitrodiphenyl
4. 2-methoxy quinoline
5. Penta-2,3-diene-1-oic acid

C) a) Explain convergent synthesis with a suitable reaction? 3

b) Define chemoselectivity with a suitable example? 2

D) a) Explain multicomponent synthesis with a suitable example? 3

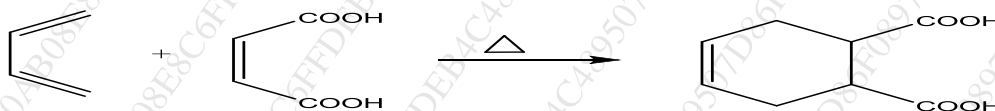
b) What is E-factor? Give its significance? 2

E) Give the synthesis of the following from a suitable starting compound. 5

1) 1-phenyl ethanol using a suitable Grignard reagent.

2) n-pentane using a suitable Organolithium compound

F) a) Define atom economy? Calculate the percentage atom economy of the following reaction? 3



[Given Atomic Weights: C=12, H=1, O=16]

b) Give any two applications of biocatalyst in green chemistry? 2

Q.4 Attempt ANY FOUR of the following:

A) a) Explain the following terms used in spectroscopy with suitable example 3

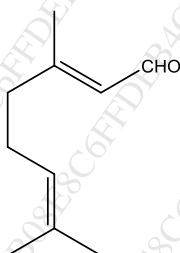
i) Chromophore ii) Auxochrome

b) Give the shifts in absorption bands in UV- visible spectroscopy. 2

B) a) Explain the fragmentation of the 2- Methyl pentane 3

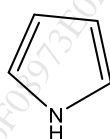
b) Define Base peak and Isotopic peak 2

C) a) What are terpenoids? Give products of ozonolysis of the following terpenoid? 3



b) State Special isoprene rule with suitable example. 2

D) Give the reaction for Hofmann exhaustive methylation and degradation of 3



b) What are harmful effects of nicotine? 2

- E) a) Give the synthesis of citral from 6- methyl hept-5-en-2-one. **3**
 b) Give analytical evidence that nicotine has pyridine ring with a side chain containing > N-CH₃ group. **2**
- F) a) Give Ott's synthesis of adrenaline? **3**
 b) Give any one analytical evidence to prove that citral is an α , β - unsaturated aldehyde? **2**
- Q.5 A) Select the correct option and complete the following statements: (ANY FIVE) **5****
- a) Base catalysed hydrolysis of esters is called
- | | |
|----------------------|-----------------------|
| i) Esterification | iii) Cope Elimination |
| ii) Chugaev reaction | iv) Saponification |
- b) Nucleophilicity is a and term.
- | | |
|-----------------------------|------------------------|
| i) relative ; thermodynamic | iii) relative; kinetic |
| ii) absolute; thermodynamic | iv) absolute; kinetic |
- c) Transition state is a characterised by partial bonds.
- | |
|---|
| i) low energy, definite molecular species |
| ii) low energy, indefinite molecular species |
| iii) high energy, definite molecular species |
| iv) high energy, indefinite molecular species |
- d) NGP assisted nucleophilic substitution reactions showin rate of reaction with of configuration.
- | | |
|-------------------------|--------------------------|
| i) decrease; retention | iii) decrease; inversion |
| ii) increase; retention | iv) increase; inversion |
- e) is a 4+2 π cycloaddition reaction.
- | | |
|----------------------|--------------------|
| i) Cope reaction | iii) Diel's Alder |
| ii) Chugaev reaction | iv) Saponification |
- f) Transfer of radiant energy from donor to acceptor molecule is called
- | | |
|------------------------|-------------------------|
| i) photoreduction | iii) photoisomerisation |
| ii) photorearrangement | iv) photosensitisation |
- g) is an allowed electronic transition.
- | | |
|-------------------------------------|--------------------------------------|
| i) S ₁ → T ₁ | iii) T ₁ → S ₁ |
| ii) S ₁ → T ₂ | iv) S ₁ → S ₂ |
- h) Photochemical cleavage of carbonyl compounds to form alkane and carbon monoxide are called reactions.
- | | |
|-----------------------|---------------------|
| i) di- π -methane | iii) Norrish Type I |
| ii) eletrocyclic | iv) Norrish Type II |

Q.5 B) State whether the following statements are TRUE or FALSE: (ANY FIVE) 5

- Trans-1,3-Dimethyl cyclobutane is chiral.
- Cumulenes having odd number of double bonds shows geometrical isomerism.
- Alternating axis of symmetry is also known as rotation reflection axis.
- Electrophilic substitution reactions on isoquinoline takes place preferably at position 5 and 8.
- Isoquinoline is also known as 2-azanaphthalene.
- Gibberelins belong to the class of Plant Growth Regulators.
- Endosulfan is a plant growth regulating hormone.

Q.5 C) Fill in the blanks with correct alternatives given in the bracket : (ANY FIVE) 5

[quinoline, chemoselective, LiAlH_4 , dimethylsulphate, larger, bridged head, smaller, isoquinoline, renewable, dimethylcarbonate, supercritical CO_2 , regioselective, tert.butyl hydroperoxide]

- Addition of halogen acid to an unsymmetrical alkene is an example of ----- reaction.
- Methane is an example of ----- raw material
- causes selective methylation of active methylene compounds
- Oxidation of aldehydes can be carried out with benign reagents like -----
- Dry cleaning of the clothes can be done using ----- instead of carbon tetrachloride.
- benzo[c]pyridine is also called -----
- For the nomenclature of spirans the ----- ring is given preference
- To name the fused and the bridged ring systems the numbering starts from one of the ----- carbon atoms.

Q.5 D) Match the columns: (ANY FIVE) 5

Column A

- Citral-b
- $>\text{N}-\text{CH}_3$
- Epinephrine
- Pinner
- Odd number of nitrogen atom
- β - carotene
- Citral-a

Column B

- Odd mass number
- Nicotine
- Hypochromic shift
- Geranial
- Laevorotatory
- Neral
- $\lambda_{\text{max}} = 452\text{nm}$
- Protein hormone
- Herzig Meyer method
