	Define	genetic		code.	Give	at	least	three
	characte	ristics	of	genetic	code.			(4.5)

- (c) Differentiate between competitive and noncompetitive enzyme inhibition and give examples of each kind.
- 7. (a) Discuss various steps involved in the glycolysis cycle, giving the names of enzymes and structures of intermediates. (6)
 - (b) List any two reactions that were not explained by the open chain structure of glucose. How does the cyclic hemiacetal structure of glucose explain them. (4)
 - (c) Define ω fatty acids giving suitable example.

(2.5)

(a) How will you convert:

- (i) D-Glucose to D- Arabinose.
- (ii) D-Glucose to D-Fructose. (4.5)
- (b) List four important characteristic of Watson and Crick model of DNA. (4)
- (c) Discuss the 2° structure of proteins. (4)

[This question paper contains 4 printed pages.]

Sr. No Act Offestion Paper: 4324

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Inique Paper Code

: 32171501

Name of the Paper

: Organic Chemistry IV:

Biomolecules

Name of the Course

: B.Sc. (Hons.) Chemistry

Semester

Duration: 3 Hours

Maximum Marks: 75

Instructions for Candidates

- 1. Write your Roll No. on the top immediately on receipt of this question paper.
- Attempt any 6 questions.
- All Questions carry equal marks.
- (a) (i) Give the sequence of reactions to establish the pyranose ring size in glucose? (3)
 - (ii) Draw Fisher projection and Haworth structure of β-D-Fructofuranose. (3)
 - (b) A pentapeptide, X with empirical composition, Phe-2, Lys, Trp, Tyr was recovered unchanged on treatment with Trypsin. DNFB treatment of X

following amino acid composition:

3

followed by hydrolysis gave DNP-Tyr. Partial acid hydrolysis of X gave two tripeptides with the

- 1. Phe, Tyr, Trp
- 2. Phe, Trp,Lys

Elucidate the sequence of amino acids in X. Give the products of treatment of X with Chymotrypsin.
(6.5)

- 2. (a) Draw the complimentary DNA sequence of the given fragment specifying the direction 5'—ATGC—3'. (4.5)
 - (b) Draw the structure of glyceryl trilinoleate and calculate its iodine value. (4)
 - (c) Fructose contains a ketonic group, yet it reduces Tollens' reagent and Fehling's solution. Why? Give the mechanism and name of reaction involved.

(4)

- 3. (a) (i) How is pyruvate converted to ethanol under anaerobic conditions? Name the enzyme and give the reaction involved. (3.5)
 - (ii) ATP is called the universal currency of cellular energy. Explain giving its hydrolytic pathway. (3)

(b) Give the complete name and structure of

(i) dAMP

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(ii) UDP (4)

(c) What are cofactors? Give an example. (2)

- 4. (a) Give the structure of the disaccharide sucrose.

 What is its IUPAC name? Account for the observation that it does not reduce Fehling's solution and it does not mutarotate. (3)
 - (b) Give the mechanism of osazone formation for D-glucose. D-Glucose and D-Fructose give the same osazone. Explain. (3.5)
 - (c) How would you synthesize the tripeptide, Ala-Gly-Ala using solid phase peptide synthesis. (6)
- 5. (a) Explain the mechanism of action of chymotrypsin. (6)
 - (b) What are drying and non-drying oils? Give one example of each type. (4)
 - (c) What is denaturation of protein? How it is caused? (2.5)
- 6. (a) An α D methylglycoside, X, with molecular formula $C_7H_{14}0_6$, consumes 2 moles of HIO_4 and gives one mole of HCHO. Arrive at possible structure/s of X. (4)

P.T.O.