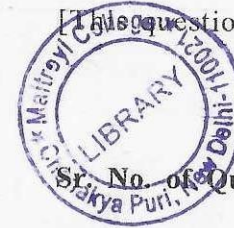


(c) Write the semi-empirical mass formula for calculation of the binding energy of a nucleus. Calculate the binding energy per nucleon for the zinc isotope  $^{64}\text{Zn}_{30}$ . The values of coefficients are :

$$a_1 = 14.1 \text{ MeV}, a_2 = 13.0 \text{ MeV}, a_3 = 0.595 \text{ MeV},$$

$$a_4 = 19.0 \text{ MeV}, a_5 = 33.5 \text{ MeV}. \quad (5,5,5)$$

[This question paper contains 8 printed pages.]



05.01.2024(M)  
Your Roll No.....

G

Unique Paper Code : 42227929

Name of the Paper : Elements of Modern Physics

Name of the Course : B.Sc. (Prog.) Physical  
Science : DSE

Semester : V

Duration : 3 Hours

Maximum Marks : 75

#### Instructions for Candidates

1. Write your Roll No. on the top immediately on receipt of this question paper.
2. Attempt **five** questions in all.
3. Question No. 1 is compulsory.
4. **All** questions carry equal marks.
5. Non-programmable scientific calculators are allowed.

1. All parts are compulsory. (3×5=15)

- (a) Monochromatic light of wavelength  $3000 \text{ \AA}$  is incident on a surface of area  $4 \text{ cm}^2$ . If the intensity of the incident light is  $15 \times 10^{-2} \text{ W/m}^2$ , determine the rate at which photons strike the surface.
- (b) Determine the wavelengths of the first two spectral lines of Lyman series of hydrogen. (Rydberg constant =  $1.097 \times 10^7 \text{ m}^{-1}$ ).
- (c) Derive the expressions for momentum and energy operators starting from the expression for the wavefunction of a free particle.
- (d) Find the lowest energy of an electron confined to move in a one-dimensional infinite potential box of length  $2 \text{ \AA}$ . (Given  $m = 9.11 \times 10^{-31} \text{ kg}$ ,  $h = 1.054 \times 10^{-34} \text{ Js}$ )
- (e) Show that the density of nuclear matter is independent of the mass number  $A$  of nuclei. Obtain an estimate of this density.

If the energy of the electron is less than the height of the barrier ( $E < V_0$ )

- (a) Write the time-independent Schrodinger equation and its solution for the two regions and explain the meaning of each term.
  - (b) Define penetration depth. Calculate the penetration depth for a  $5 \text{ eV}$  electron approaching a step barrier of  $10 \text{ eV}$ .
  - (c) A particle of energy  $E$  is incident on a potential step of height  $V_0 = 10 \text{ eV}$ . Calculate the reflection coefficient  $R$  and the transmission coefficient  $T$  when  $E = 5 \text{ eV}$  and  $15 \text{ eV}$ . (5,5,5)
7. (a) Discuss Gamow's theory of  $\alpha$ -decay.
- (b) Find the energy released, when two  ${}^2\text{H}_1$  nuclei fuses together to form  ${}^4\text{He}_2$  nucleus. The binding energy per nucleon of  ${}^2\text{H}_1$  and  ${}^4\text{He}_2$  is  $1.1 \text{ MeV}$  and  $7.0 \text{ MeV}$ , respectively.

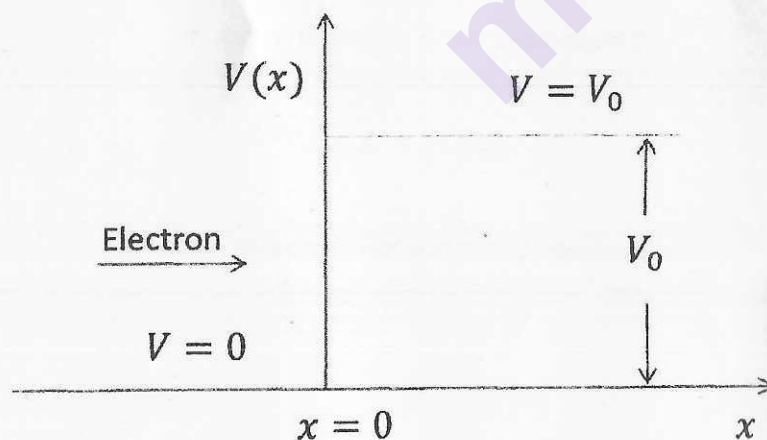
- (c) Prove the following relation for the case of a particle undergoing relativistic motion :

$$\lambda = \frac{hc}{\sqrt{K(K + 2m_0c^2)}}$$

The symbols have their usual meaning. (5,5,5)

6. An electron of energy 'E' approaches a step barrier potential of height  $V_0$  from the left as shown.

$$\text{Here, } V(x) = \begin{cases} 0 & \text{for } -\infty < x \leq 0 \\ V_0 & \text{for } 0 \leq x \leq \infty \end{cases}$$



2. (a) Which aspects of the photoelectric effect cannot be explained by classical physics?

- (b) Consider a photon that scatters from an electron at rest. If the Compton wavelength shift is observed to be triple the wavelength of the incident photon and if the photon scatters at  $60^\circ$ , calculate the wavelength of the incident photon.

- (c) State the postulates of Bohr's atomic model. Derive the expressions for the radius of  $n^{\text{th}}$  Bohr orbit and the velocity of electron in the  $n^{\text{th}}$  orbit.

(5,5,5)

3. (a) A beam of electrons is incident normally on a double slit. Draw the expected and observed patterns on the screen when both slits are open. How does the linear superposition principle help in understanding the observed pattern?



- (b) An electron has the speed of  $600 \text{ ms}^{-1}$  with an accuracy of 0.005%. Calculate the minimum uncertainty in determining its location.
- (c) Calculate the time required for 10% of a sample of thorium to disintegrate. Take the half-life of thorium to be  $1.4 \times 10^{10}$  years. (5,5,5)
4. (a) An object's one-dimensional motion is described by the wavefunction

$$\psi(x) = \begin{cases} Ax, & 0 < x < 1 \\ 0, & \text{elsewhere,} \end{cases}$$

where A is the normalization constant. Determine A and calculate the probability of finding the particle in the region  $x = 0$  to  $x = 0.5$ ?

- (b) Using Heisenberg's uncertainty principle find an estimate for the ground state energy of an electron confined in an infinite potential well of width L.

- (c) Assume that the electron in a hydrogen atom is replaced by a muon of mass  $m \approx 200 m_e$ , where  $m_e$  is the mass of an electron. Determine the change in the wavelength of the  $H_\alpha$  line of Balmer series. Assume  $m \ll m_p$ , where  $m_p$  is the mass of a proton. (5,5,5)
5. (a) State the conditions for a physically acceptable wavefunction. Discuss about its physical significance.
- (b) Define probability current density. For a 1-dimensional wave function given by

$$\psi(x, t) = A(x)e^{i\phi(x,t)}$$

evaluate the probability current density.