[Time: 2.30 Hours]

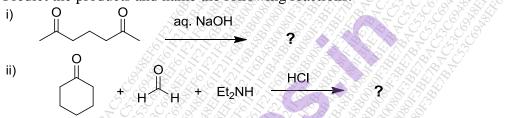
[Total Marks: 60]

Please check whether you have got the right question paper.

NB: 1. All questions are compulsory.

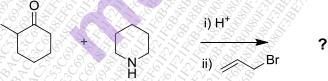
- 2. **Figures** to the **right** indicate **full** marks.
- Q.1(A) Attempt any **two** of the following:
 - (a) Write the major product A and B with explanation of its formation under given conditions.

(b) Predict the products and name the following reactions.



(c) Predict the products and give the mechanism of the following reaction.

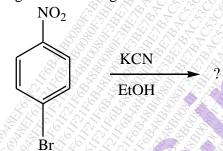
- (d) What are enamines? Explain with suitable example alkylation of enamines. 4
- (B) Attempt any **one** of the following:
- (a) Complete the following reaction with its mechanism.



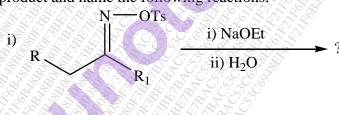
- (b) Explain Knoevenagel amine catalysed condensation and give its mechanism. 4
- Q.2(A) Attempt any **two** of the following:
 - (a) What is Baylis-Hilman reaction? Explain its mechanism.
 - (b) Explain Rupe rarrangement with its mechanism.
 - (c) Predict the product of the following reactions and give the name of the reactions.

i)
$$\begin{array}{c} & & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\$$

(d) Complete the following reaction and give its mechanism.



- (B) Attempt any **one** of the following:
- (a) Explain Lossen rearrangement with its mechanism.
- (b) Predict the product and name the following reactions.

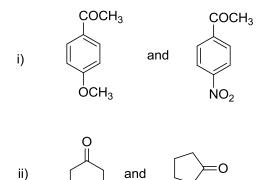


ii)
$$H_3C$$
 CH_3 $PhLi$

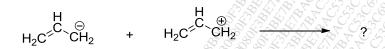
4

- Q.3(A) Attempt any two of the following:
 - (a) i) Explain the importance of the fingerprint region in IR Spectroscopy?
 ii) Explain λmax of 1,3 butadiene is 217 nm while that of 1,3,5-hexatriene is 254 nm
 - (b) Explain the addition of hydride to formaldehyde with the help of FMOs. 4
 - Which of the following compounds will show a lower frequency in IR spectroscopy and why?

Paper / Subject Code: 92152 / Chemistry: Paper III - Organic Chemistry



(d) Complete the following reaction. Draw the π MOs diagram for the reactants. 4 Show the interaction of their FMOs.



- (B) Attempt any **one** of the following:
- (a) Draw the π MOs of a diene and dienophile showing electron distribution. 4 Label the FMO show their interaction in Diels-Alder reaction.
- (b) Using IR Spectroscopy how can you distinguish between 4
 - i) Maleic acid and Fumeric acid
 - ii) o-hydroxy benzoic acid and m-hydroxy benzoic acid

Q.4(A) Attempt any **two** of the following:

(a) An organic compound with M.F. C₄H₈O₂ having two isomers as **A** and **B** with I.R. stretching frequency 1745 cm⁻¹ and 2960 cm⁻¹ showed following NMR spectral data;

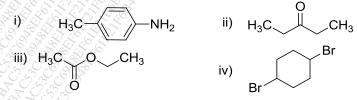
Isomer $\hat{\mathbf{A}} = \delta 0.9$ (3H, t, J = 7.5 Hz), $\delta 4.5$ (2H, q, J = 7.5 Hz), and $\delta 2.3$ (3H, s).

Isomer **B** = δ 0.9 (3H, t, J = 7.5 Hz), δ 2.4 (2H, q, J = 7.5 Hz), and δ 4.3 (3H, s).

Deduce the structure of A and B isomer and justify your answer.

- (b) Write the fragmentation pattern of the following molecules in Mass spectroscopy.
 - i) Benzoic acid ii) Ethyl acetate
- (c) Explain the following terms in NMR spectroscopy. 4
 - i) First Order spectra ii) Karplus equation
- (d) Explain the following in Mass spectroscopy with suitable example.i) Nitrogen ruleii) Retro Diels Alder reaction
- (B) Attempt any **one** of the following:
- (a) Write the number of signals in proton decoupled and off resonance proton decoupled ¹³CNMR spectra of following molecules:

4



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Page 3 of 4

(b) An Organic Compound having M.F. C₆H₁₃ON showed following spectral dada:

4

3

3

3

I.R. = 3500 cm^{-1} , 2970 cm^{-1} , 1690 cm^{-1} , 1050 cm^{-1} $^{1}\text{H-NMR} = \delta \ 0.9 \ (6\text{H, d,})$, $\delta \ 1.3 \ (1\text{H, m})$, $\delta \ 2.4 \ (2\text{H, d})$, $2.9 \ (3\text{H, s})$ and $\delta \ 3.0 \ (1\text{H, bs}, D_{2}O \text{ exchange})$.

Deduce the structure of the compound and justify your answer.

- Q.5 Attempt any **four** of the following:
 - A Explain how solvent polarity affects the rate of alkylation of enolates.
 - B Predict the products and give the mechanism of the following reaction.



- C Explain Tiffeneau-Demjanov rearrangement with a suitable example.
- D Complete the following reactions.

i)
$$CH_3CN + Me$$
 + C_6H_5COOH ?

ii)
$$CH_3CH_2CH_2NO_2 \xrightarrow{i) NaOH}$$
?

iii)
$$Me$$
 $TiCl_4$ $Zn(Cu)$?

- E What are hard and soft reactive sites? Give one example of each.
- F Calculate the λ max of the following compounds

- G How will you distinguish between 2-pentanone and 3-pentanone by Mass Spectroscopy?
- H Explain the magnetic anisotropy with example as benzaldehyde.