## CHE102 - Chemistry : Paper-II (Organic Chemistry) P. Pages: 2 GUG/W/18/1203 Time : Three Hours Max. Marks : 50 All questions are compulsory and carry equal marks. Notes : 1. 2. Draw chemical equation wherever necessary. 1. a) What is fission of covalent bonds? Explain the types of fission of covalent bond with 5 example. Explain the following reaction with examples. 5 b) Elimination reaction Rearrangement reaction ii) i) OR $2^{1/2}$ Discuss Inductive effect. c) Explain hybridization in ethylene. $2^{1/2}$ d) $2^{1/2}$ What are carbocations? Explain stability of carbocations. e) Explain substitution reaction with example. $2^{1/2}$ f) 2. Explain optical activity of Tartaric acid. 5 a) What is resolution? Explain its different methods of resolution. 5 b) OR Explain geometrical isomerism of maleic and fumaric acid. $2^{1/2}$ c) Discuss conformation of ethane. $2^{1/2}$ d) What is isomerism? Give its different types. $2^{1/2}$ e) f) Discuss sequence rule for R-S nomenclature. $2^{1/2}$ 3. Explain Bayer's strain theory with its draw backs? 5 a) 5 b) Give preparation of cycloalkane by-Freund's method i) ii) Dickmann's method OR What are dienes? Give its classification. $2^{1/2}$ c) Explain oxidation of cycloalkane with example. $2^{1/2}$ d) Explain Kolbe's synthesis for preparation of alkane. $2^{1/2}$ e) Explain polymerization of alkene with example. $2^{1/2}$ f)

Bachelor of Science (B.Sc.) (Part-I) First Semester (Old)

## GUG/W/18/1203

4.	a)	Explain structure and orbital diagram of benzene.	5
	b)	What is orientation? Discuss O, P orientating effect in – OH group in phenol.	5
		OR	
	c)	Explain Huckel's rule of aromaticity.	21/2
	d)	Discuss Birch reduction with example.	21/2
	e)	Explain activating and deactivating groups with example.	21/2
	f)	Explain nitration in benzene.	21/2
5.		Attempt any ten.	1x10=10
		1) Write bond angle in SP and $SP^2$ hybridization.	
		2) Define free radicals.	
		3) Define electromeric effect.	
		4) Draw Newman projection formula for ethane.	
		5) Define Asymmetric synthesis.	
		6) Identify cis & trans forms of following.	
		$CH_{3} C = C H CH_{3} C = C H_{3}$ $H I CH_{3} H H H H$ $H I CH_{3} C = C H_{3}$ $H I I H$	
		7) Define octane number.	
		8) Explain oxidation of ethene in presence of mild oxidizing agent.	
		9) State Markownikoff's rule.	
		10) What is Friedal-craft alkylation?	

- 11) Identify O, P orientating groups in the following -OH,  $-NO_2$ .
- 12) Which compound of following is aromatic?

