NOTE: i) All the questions are compulsory.

- ii) Figures to right indicate full marks.
- iii) Use of non -programmable calculator / log table is allowed.

Q.1. Attempt any four:

[20]

- A) Explain Nitration of Benzene giving its reaction , mechanism and energy profile diagram.
- B) Give 5 applications of Nitro compounds.
- C) Write a note on Sandmeyer's reaction.
- D) Give IUPAC nomenclature of following compounds -

- E) Write a note on Knoevengel Reaction giving its mechanism.
- F) Give synthesis of

(1)Orange II

(2) Methyl orange

- G) Write 2 advantages of H₂SO₄ in Nitration reaction and Give Synthesis of Congo red.
- H) Explain Rosenmund Reduction of Aroyl halide with its mechanism.

Q.2. Attempt any four:

[20]

- A) Give reaction and mechanism of Sulphonation of benzene
- B) Explain conversion of -
 - (1) Bromobenzene to benzoic acid by Grignard's reagent.
 - (2) Ethyl benzene to Benzoic acid.
- C) Explain Kolbe Schmidt Reaction.

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D) Write synthesis of following compounds -

- 1) m-nitro benzoic acid
- 2) 4-Amino-2-bromo toluene
- E) How is Benzoic acid prepared from-Aryl nitriles
- F) Draw the conformations of n-butane and discuss their relative stabilities.
- G) Complete the following reactions-

(i)
$$O = C - OH + HOCH_3 = \frac{H_250H}{2}$$
? + H_2O

H) (1) Assign R,S descriptors to the following compounds -

(i) CHO (ii) CH₃ (iii) COOH
$$H - C - CI$$
 $CI - C - CH2(I)$ $CH3 $- C - NH2$ $CH2(I)$ $CH2(I)$ $CH2(I) $CH3(I)$ $CH$$$

(2) Assign E, Z descriptors to the following compounds-

$$(i) cH_3 = (ii) cH_3 = (iii) cH_3 = (iii)$$