

Note: i) All the questions are compulsory.

ii) Numbers indicate in the right hand side are marks

iii) Use of non programmable calculator is allowed.

Q.1A) Answer the following questions.(ANY 1)

(10M)

- Derive an expression for efficiency of otto engine.
- Derive an expression for work done in a carnot's cycle.

Q.1B) Answer the following questions.(ANY2)

(10M)

- For a reversible engine ,show that $\sum \frac{Q}{T} = 0$.
- Show that all reversible engines working between the same two temperatures have the same efficiency.
- State various statements of the second law of thermodynamics.
- Find the efficiency of Carnot engine working between 127°C and 27°C . It absorb 80 calories of heat. How much heat is rejected.

Q.2A) Answer the following questions.(ANY 1)

(10M)

- Derive an expression for the change in entropy of perfect gas in terms of temperature and Volume.
- If 100 grams of water at 0°C is mixed with an equal mass of water at 86°C , Calculate the resultant increase in entropy.

Q.2B) Answer the following questions.(ANY2)

(10M)

- What is the change in entropy, if 10 gm of water is heated from 313 K to 353K?
- Write a short note on Entropy and disorder.
- Define Entropy. Explain Entropy remains constant of a closed system during reversible process.
- Write a short note on heat death of universe.

Q.3A) Answer the following questions.(ANY 1)

(10M)

- What is thermocouple? Explain different type of thermocouple in detail with their neat and labelled diagram.
- Explain in detail Semiconductor Diode Temperature Sensor.

Q.3B) Answer the following questions.(ANY2)

(10M)

- With neat and labelled diagram explain construction of resistance thermometer.
- Distinguish between thermistor and thermocouple.
- Write short note on Ultrasonic temperature transducer.
- Write short note on Platinum thin film sensor.

Q.4) Answer the following questions.(ANY3)

(15M)

- A 10 gm of water is heated from 40°C to 80°C . Calculate the change in entropy.
- An ideal heat engine operates according to Carnot's cycle and receives 600 cal from heat source every cycle. The temperature of the heat source is 400 K and that of the sink 300K. Find the work done by the Carnot cycle and amount of heat rejected to the sink for the cycle.
- Give advantages and disadvantages of RTD.
- What is pyrometer? Describe, in brief, a disappearing filament type of optical pyrometer.
- Give Comparison of Otto engine and diesel engine.
- What is Reversible and Irreversible Processes?

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