27/11/19

VCD F.Y.B.Sc SEM-I Regular+ATKT PHYSICS-I 2019-20 100 MARKS 3 HRS

Note: (1) Al	l questions	are compul	sory.
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- (2) Figures to the right indicate maximum marks.
- (3) Use of non-programmable calculators is permitted.
- (4) Symbols used have their usual meaning.

Q.1. (A) Choose correct alternative in each of the following: (12)

- (i) In Ramsden's eyepiece, if the focal length of eye lens is 12cm, then its equivalent focal length is
 - a) 12cm b) 9cm c) 6cm d) 24cm
- (ii) Two plano-convex lenses of same refractive index form a Huygen's eyepiece. Their focal lengths are 6cm and 2cm. what is the equivalent focal length of the eyepiece?
 - a) 2cm b) 6cm c) 3cm d) 4cm
 - (iii) For a body undergoing acceleration in an inertial frame, the net force acting on the body is
 - a) Zero b) non-zero c) balanced by pseudo force d) proportional to the momentum of the body
 - (iv) A person standing in an elevator feels an increase in his weight when the elevator is
 - a) Ascending with retardation
- b) ascending with acceleration
- c) descending with acceleration
- d) stationary
- (v) The Boyle's temperature of a gas is given by
 - a) a/bR b) a/27b²
 - b) a/27b² c) 8a/27bR d) a/27

Where symbols have their usual meanings.

- (vi) The dimensions of the constant 'b' in Van der Waals equation of a state of a gas are
 - a) Volume b) pressure c) pressure/volume d) volume/pressure
- Q.1. (B) Answer in one statement: (03)
 - (i) Define focal length of a lens.
 - (ii) What is an inertial frame of reference?
 - (iii) What is an isochoric process?

(ii) Determine the radii of curvature for a lens of focal length f = 30cm and refractive index 1.5

such that parallel incident light has minimum spherical aberration.

- Q.4. (A) Attempt any one: (08)
- (i) Explain the behavior of gases at high pressure and obtain the expression for the Boyle's temperature.
- (ii) Derive an expression for the work done in adiabatic and isothermal processes.
- Q.4. (B) Attempt any one: (08)
- (i) Discuss Andrew's experiment on carbon dioxide and explain the result using P-V diagram.
- (ii) Derive the relation between the Van der Waals constants and critical constants. State the first law of thermodynamics and hence what is its significance and limitations.
- Q.4. (C) Attempt any one: (04)
- (i) A quantity of air at 27° C and atmospheric pressure is suddenly compressed to 25% of its original volume. Find the final pressure and temperature if ratio of specific heat (C_p/C_V) is 1.4.
- (ii) Calculate the value of the constants a and b in Van der Waals equation for dry air, given that for dry air, $T_C = 132 \text{ K}$, $P_C = 37.2 \text{ atm}$, $R = 82.07 \text{ cm}^3 \text{ atmosphere K}^{-1} \text{ mole}^{-1}$.
- Q.5. Attempt any four: (20)
 - (i) Explain how spherical aberration can be reduced using stops or by reducing the lens aperture.
 - (ii) Derive a general expression for $C_P C_V$.
 - (iii) State and explain zeroth law of thermodynamics.
 - (iv) Write a brief notes on chromatic aberration.
 - (v) Define viscosity. Hence explain streamline and turbulent flow of a liquid.
 - (vi) Air at N.T.P. is compressed adiabatically to half of its volume. Calculate the change in its temperature.