

[Time: Three Hours]

[Marks: 100]

- N. B.:** (1) All questions are **compulsory**.
 (2) **Figures** to the **right** indicate **full** marks.
 (3) Draw **neat** diagrams wherever **necessary**.
 (5) Symbols have usual meaning unless otherwise stated.
 (5) Use of **non-programmable** calculator is allowed.

- Q.1. A) Select the correct alternative** **12**
- (i) 10 g of radioactive material of half-life 15 years is kept in store for 20 years. The disintegrated material is _____
 a. 9.5 g b. 3.96 g
 c. 6.04 g d. 4.03 g
- (ii) The sodium nucleus ${}_{11}\text{Na}^{23}$ contains _____
 a. 11 electrons b. 12 protons
 c. 23 protons d. 12 neutrons
- (iii) The minimum energy required for pair production is _____
 a. 0.511 keV b. 1.022 MeV
 c. 0.511 MeV d. 5.11 keV
- (iv) The average energy needed to ionize gas (eg. air) is of the order of a few _____
 a. keV b. MeV
 c. tens of eV d. hundreds of eV
- (v) Wavelength of matter waves is _____
 a. $\lambda = \frac{p}{hc}$ b. $\lambda = \frac{pc}{h}$
 c. $\lambda = \frac{p}{h}$ d. $\lambda = h/p$
- (vi) In Compton effect, the wavelength of the scattered ray is always _____ that of the incident ray.
 a. greater than b. less than
 c. greater than or equal to d. less than or equal to
- B) Answer in one sentence** **3**
- (i) What are isotones?
 (ii) Define nuclear fusion.
 (iii) How can we control the penetration power of X-rays?
- C) Fill in the blanks** **5**
- (i) A radioactive nucleus ${}_{92}\text{X}^{235}$ decays to ${}_{91}\text{Y}^{231}$. Radiations emitted are _____
 (ii) Half-life of ${}^{210}\text{Bi}$ is 5 days. If we start with 50,000 atom of this isotope, the number of atoms left over after 10 days will be _____
 (iii) If the Q value of a nuclear reaction is positive then the reaction is termed as _____

- (iv) A nuclear reaction in which the projectile picks up a nucleon from the target is termed as _____ reaction.
- (v) The gain in energy of a photon that falls in a gravitational field is manifested as increase in _____

Q. 2 A)

Attempt ANY ONE

8

- (i) On the basis of Rutherford's alpha scattering experiment, how will you estimate the size of the nucleus?
- (ii) State the law of successive disintegration of radioactive substance. Explain transient equilibrium.

B)

Attempt ANY ONE

8

- (i) Define and explain binding of a nucleus. Sketch the graph of binding energy per nucleon against mass number. Explain its characteristic features.
- (ii) State and explain law of radioactive decay. Hence obtain the expression for half-life period.

C)

Attempt ANY ONE

4

- (i) Find the distance of closest approach, when alpha particles of energy 5.48 MeV are bombarded on ^{79}Au .
- (ii) In natural carbon the abundance of ^{14}C is 1.3×10^{-12} and 5730 years is its half-life time. Calculate the number of disintegration/hour in 1 gram of natural carbon.

Q. 3 A)

Attempt ANY ONE

8

- (i) Using the concept of compound nucleus formation, derive an expression for the threshold energy of a nuclear reaction.
- (ii) Explain the construction and working of a GM counter.

B)

Attempt ANY ONE

8

- (i) Derive an expression for Geiger rule i.e. $V_0^3 = CR$ where V_0 is initial velocity of the emitted particle and R is the range of the particle and C is a constant for the particular medium in which the range is defined.
- (ii) Derive an expression for the nuclear disintegration energy i.e. balance of mass and energy in a nuclear reaction (Q-value)

C)

Attempt ANY ONE

4

- (i) Calculate the Q value for the reaction: $Pb^{210}(Fe^{54}, Fe^{56})Pb^{208}$
Given: The masses of $Pb^{208} = 207.976641$ amu,
 $Fe^{56} = 55.934939$ amu, $Fe^{54} = 53.939612$ amu,
 $Pb^{210} = 209.984178$ amu, $1 \text{ amu} = 931.5 \text{ MeV}$
- (ii) A neutron beam is incident on a stationary target of F^{19} atoms. The reaction $F^{19}(n, p)O^{19}$ has a Q-value of -3.9 MeV. Calculate the lowest neutron energy that will make this reaction possible. Assume the masses of nuclei are equal to their mass numbers expressed in "amu". Given: $1 \text{ amu} = 931.5 \text{ MeV}$

- Q. 4** **A)** **Attempt ANY ONE** **8**
- (i) Describe Laue's experiment on X-ray diffraction.
- (ii) What is Compton effect? Write the experimental determination of the Compton shift.
- B)** **Attempt ANY ONE** **8**
- (i) Give the elementary proof of Heisenberg's Uncertainty Principle.
- (ii) Describe continuous and characteristic X-ray Spectra.
- C)** **Attempt ANY ONE** **4**
- (i) A bullet of mass 20 gm is moving with a speed of 350 m/s, measured with accuracy of 0.05%. Calculate the uncertainty in the location of the bullet.
- (ii) X-ray tube emits X-rays with minimum wavelength 0.1 \AA . What is the operating voltage of the tube?
- Q. 5** **Attempt ANY FOUR** **20**
- (i) The number of first daughter element is given by; $N_2 = \frac{\lambda_1 N_0}{\lambda_2 - \lambda_1} [e^{-\lambda_1 t} - e^{-\lambda_2 t}]$. Estimate the time (t_m) taken by second daughter to attend maximum.
- (ii) Write a short note on nuclear charge and nuclear density.
- (iii) Find the Q-value for the following reaction $N^{14}(\alpha, p)O^{17}$. The masses of the nuclei are given as mass of nitrogen = 14.00753 amu, mass of oxygen = 17.0045 amu, mass of alpha particle = 4.00387 amu and mass of proton is 1.00814 amu. Given: 1 amu = 931.5 MeV
- (iv) Write short note on nuclear fission and nuclear fusion.
- (v) Derive Bragg's equation for crystals.
- (vi) In Compton scattering experiment, X-rays are scattered at angle 60° with respect to the incident beam. If the wavelength of scattered X-ray is 3.022 \AA , calculate the wavelength of incident rays.
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