Sem II 2015 ~16

QP Code: 12764

					(21/2 Hou	rs)		[Total M:	arks :75
		(1) A	Il questions a	re compulse	ory.				
N	в.:	(a) In	ternal choic	es have bee	n given.				
		(2)	gures to the	right indica	ite full ma	ks.			
		(-)	,						
		mill in	the blanks(an	ny three):-					3 , C
1.	(A)	(i)	Glycolysis	in the anaero	bic conditio	ns leads to		producti	ion.
			(a) Pyruva		b) Lactate			Malate	O^*
		°(ii)	HMP shun	,	,				<u>,</u>
			(a) Cytoso	01 (b) Mitoch	ondria	(c)	Golgi complex	7,
		(iii)	(a) 0) 10a	is consid	lered to pla	y a catalyti	ic role	in citric aciday	cle.
			(a) Succir	ic acid (b) a-Keto	glutarate	(c)	Oxaloacetale	1,4
				enesis is re		J		· P	
		(iv)	(a) Gluca	,	b) Lactate		(c)	Both's and 'b'	
			Olympagn :	nhaenharula	ee ie a kev	enzyme i	1	14.	
		(v)	(a) Gluco	neogenesis(h) Glycos	enesis.	(c)	Hycogenolysis	
			Pyruvate is	converted	to acetyl (oA by	, S.		
		(vi)	(a) Oxida	tiva (b) Reduc	tive	Br	Both 'a' and 'b'	
			(a) Oxiua	avulation (decarb	oxylation	8	- coatM	
			decard	(any analy	decaro				2
	(B)	Defin	e and explain	s converted tive (poxylation (any one):- genesis		CoA			
		(i)	THE TAX AND ADDRESS OF THE PARTY OF THE PART			ME			
		(ii)	Glycogene	SIS	0	, v			4
	(C)	Write	short notes o	n (any one)	XP				
		(i)	Glyoxylate	e pathway	81	0.4			
·	TO A	(ii)	Oxidation	of pyruvate	to acetyl	CoA			6
. 4	(D)	71 1 -	rate on (any	one):-	O				U
	(D)	(i)	TCA with	reactions	\				
		(ii)	HMP shun	t with react	ions				
		(11)		Thy.				State of the state	
	(4)	Dill in	the blanks(a	nythree):-					3
2.	(A)			Shuttle us	ed by skele	tal muscle	s and	brain is	 \
		(i)	(a) May St	e-Aspartate	shuttle			1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
			(a) Iviata	rol phospha	ate shuttle	. I risk	1		
			(b) Giyee	lol phospin	2.00				
			(c) Both	a and b	a the dene	ndence o	f ele	ctron transfer o	n ATP
	(ii) explains the dependence of electron transfer on								
		^	. synthesis i	n mitochon	aria.				
		'M	(a) Proton	n-motive fo	rce				
		45	(b) Rotat	ional cataly	sis				
	17	7	(c) Chem	iosmotic th	eory				(Mai)
	N		(0) 0.23					[TURN O	VER
	KI								

(A) Fill in the blanks(any three):-

The stationary phase in reverse phase partition chromatography is supported by a matrix.

(a) Non-polar compound

(b) Polar compound

Water

(a) Rf

(b) Kd

(c) aWr

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3/	CM	-Con	. 2228-15.	
01,3)			[TURN OVER
5	Ta,		(c) 9.74	
	1		(b) 2.34	
	, F	-	(a) 5.97	
		(iii)	pI value of lysine is	
		C/X	(c) Pyridine	
		3	(b) Chromane	
4				THE RESERVE TO SERVE THE PARTY OF THE PARTY
		(ii)	Rixoflavin consists of sugar alcohol attached to	ni taya ay iya k
			(c) Behydrocholesterol	
			(b) Cholecalciferol	Carried St.
		h Baile	(a) Ergocalciferol	nighter with the state of the
		(i)	The active form of vitamin D in humans is	
4.	(A)		n the blanks(any three):-	3
				an ing a sample of the sample
		(ii)	Paper chromatography	
		(i)	Adsorption chromatography	
	(D)		rate on (any one):-	6
		(ii)	Application of gel filtration shromatography	September 1985
	(0)	(i)	Principle of HPLC	
	(C)	Write	short notes on (any one):-	A Comment of the Comm
		(ii)	Ion exchangers	
	(D)	(i)	Stationary phase	2
	(B)	Defin	(c) Both a and b If Kd is, moderately moving zones at (a) Greater than 1 (b) 1 (c) Less than 1 e and explain (any one):- Stationary phase Ion exchangers short notes on (any one):- Principle of HPLC Application of gel filtration shromatography rate on (any one):-	
			(c) Less than 1	
			(b) 1	
	4.57	(vi)	(a) Greater than 1	o obtained.
		(in)	If Kd is moderately moving zones	A obtained
			(c) Both 'a' and 'b'	The
			(a) Polar (b) Non-polar	AAA
		(v)	In TLC, for charged lipids,solvents as (a) Polar	re used. (X)
			(c) H ₂ SO ₄ in ethanol	re used the
			(b) Ninhydrin	St
			(a) Iodine vapour	\mathcal{O}
			compounds are being investigated.	
		(iv)	In the detection of component, is use	eful if unsaturated
			(c) Resin	
			(b) Sephadex	
			(a) Celite	
			on	
		(iii)	In Obe, a successful phase of smoon great	se is supported

		(iv	The Isoelectric pH of glycine is calculated by arithmetic mean of pK1 and	
			(a) pK	
			(b) pK,	
		grilly.	(6) pK	
		(v)	takes part as coenzyme in transamination. (a) Pyridoxine (b) Flavin (c) Biotin In titration curve, the relatively flat zone observed is called (a) Titration zone (b) Acceptor zone (c) Buffering zone and Explain the following(any one):- Vitamins FAD short notes on (any one):- Role of vitamin A in vision Coenzyme form of vitamin B1 ss in detail(any one):- Working of glass electrode and its applications Titration curve of asparate (any one of the following:- Citric acid cycle is amphibolication of the following:- Citric acid cycl	ئے
			(a) Pyridoxine	6
			(b) Flavin	7
		(vi)	(c) Biotin	
		(11)	(a) Titration zone	
			(b) Acceptor zone	
			(c) Buffering zone	
	(B)	Defin	ne and Explain the following(any one):-	2
		(1)	Vitamins	
	1.00	(ii)	FAD 2	
	(C)	440	short notes on (any one):-	4
		(i)	Role of vitamin A in vision	
	(6)	(ii)	Coenzyme form of vitamin B1	
	(D)		ss in detail(any one):-	6
		(i) (ii)	Working of glass electrode and its apparent	
		(11)	Tittation curve of asparate	
5,	(A)	Justif	any one of the following:-	2
	(-)	(i)	Citric acid cycle is amphibolic	-
		(ii)	Glycolysis in humans is highly significant.	
	(B)		ate on (any one):-	3
	. 100	(i)	Uncouplers of ETC	
		(ii)	Electron carriers in Anotosynthesis	
	(C)	Explain	n any one of the following:-	3
		(i)	Application of ion exchange chromatography	
		(ii)	Principle of affinity chromatography	3
	(D) .		at any one of the following:-	•
		(i)	Physiological role of vitamin E	
		(ii)	Calculate the pH of a mixture of 0.1M acetic acid and 0.2M sodium	
	root in	/ 2/2//	acetate $(pK_n = 4.76)$.	3
1	(E) \		rue or False(any three):- Plyceraldehyde-3-phophate is formed at the end of phase I of glycolysis.	
		(i) _	Transketolases form part of gluconeogenesis.	
		(11)	Ubiquinone is also known as CoQ10.	
		(HO)	Affinity chromatography relies on differences in the physical properties	
	X	(rv)	of the molecules to be separation of buffers.	
	F	10	Henderson-Haselbalch equation is used for preparation of buffers.	
1	1	(v)	Riboflavin and Lactoflavin are different biomolecules.	
V	• ((vi)	KIDOHAVIII aliu Lactonaviii ale uniterent etemper	-
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